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you learned in Chapter 4,
Rutherford proposed that all of an
atom's positive charge and
virtually all of its mass are
concentrated in a nucleus that is
surrounded by fast-moving
electrons. Although his nuclear
model was a major scientific
development, it lacked detail
about how electrons occupy the
space surrounding the nucleus.
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compare the wave and particle
models of light. You will describe
how the frequency of light ...
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speed of all electromagnetic
waves in a vacuum is the same,
waves can have different
wavelengths and frequencies.
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answer key. Atoms are made of
extremely tiny particles called
protons, neutrons, and electrons.

Protons and neutrons are in the
center of the atom, making up the
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